

*Lectures and practical exercises*

<b>Subject:</b>	<b>Medical Chemistry</b>	<b>Code:</b>	<i>ULCHBKB/MCH-DM/22</i>
<b>Study Programme:</b>	<i>Dental Medicine</i>	<b>Study Period:</b>	<i>1. semester</i>
<b>Evaluation:</b>	<i>Exam</i>	<b>Subject Type:</b>	<i>compulsory</i>
<b>Content:</b>	<i>2 h lectures and 3 h practical exercises / week</i>		<i>Total 70 hours</i>

Workplace: **Department of Medical and Clinical Biochemistry, UPJŠ in Košice, FM**

<i>Week</i>	<i>Lectures</i> <a href="https://portal.lf.upjs.sk/index-en.php">https://portal.lf.upjs.sk/index-en.php</a>	<i>Practical Lessons</i> <a href="https://portal.lf.upjs.sk/index-en.php">https://portal.lf.upjs.sk/index-en.php</a>
1.	<p><b>INTRODUCTION TO MEDICAL CHEMISTRY</b></p> <ul style="list-style-type: none"> <li>- Definition of terms</li> <li>- International (English) nomenclature</li> <li>- Properties of elements, micro- and macro-elements</li> </ul> <p><b>WATER AND SOLUTIONS</b></p> <ul style="list-style-type: none"> <li>- Water, properties and biological function of water</li> <li>- Disperse systems, properties of disperse systems</li> <li>- Solutions</li> </ul>	<p><b>PRINCIPLES OF LABORATORY TECHNIQUE</b></p> <ul style="list-style-type: none"> <li>- Laboratory safety rules</li> <li>- Equipment of laboratory bench</li> <li>- exercise: <i>Volume measurement</i></li> </ul>
2.	<p><b>BIOCHEMICAL REACTIONS PROCESS RULES I</b></p> <ul style="list-style-type: none"> <li>- Kinetics of chemical reaction</li> <li>- Effect of reactant concentration, temperature on reaction rate. Order of reaction rate</li> <li>- Arrhenius, Brønsted-Lowry and Lewis theory</li> <li>- Autoprotolysis, H<sup>+</sup> exponent</li> <li>- pH of acids and bases. Buffer systems. Hydrolysis of salts</li> <li>- Equilibrium of a chemical reaction, equilibrium constant</li> <li>- Impact of factors on chemical equilibrium</li> </ul>	<p><b>NOMENCLATURE OF INORGANIC COMPOUNDS (English)</b></p> <p><b>CALCULATIONS</b></p> <ul style="list-style-type: none"> <li>- Stoichiometric calculations</li> </ul>
3.	<p><b>BIOCHEMICAL REACTIONS PROCESS RULES II</b></p> <ul style="list-style-type: none"> <li>- The basics of chemical thermodynamics</li> <li>- Internal energy, enthalpy, entropy</li> <li>- Gibbs free energy, chemical reaction spontaneity</li> <li>- Precipitation reactions, solubility equilibrium</li> <li>- Complexation reactions, complex formation Redox reactions, redox potential</li> <li>- Electrode potential of metals, Electrochemistry</li> </ul>	<p><b>CALCULATIONS</b></p> <ul style="list-style-type: none"> <li>- Solutions</li> </ul> <p><b>VOLUMETRIC ANALYSIS</b></p> <ul style="list-style-type: none"> <li>- exercise: <i>Standardization of NaOH solution</i></li> </ul>
4.	<p><b>ORGANIC CHEMISTRY I</b></p> <ul style="list-style-type: none"> <li>- Alcohols and aldehydes</li> <li>- Carboxylic acids</li> <li>- Functional derivatives of carboxylic acids</li> <li>- Substitutional derivatives of carboxylic acids</li> <li>- Carbonic acid derivatives</li> </ul>	<p><b>CALCULATIONS</b></p> <ul style="list-style-type: none"> <li>- Calculation of pH of acid, base, and salt solutions</li> </ul> <p><b>pH MEASUREMENT</b></p> <ul style="list-style-type: none"> <li>- exercise: <i>Preparation and pH measurement of acetate buffers</i></li> </ul>
5.	<p><b>ORGANIC CHEMISTRY II</b></p> <ul style="list-style-type: none"> <li>- Organic sulphur, phosphorus, nitrogen compounds, structure, and biological significance</li> <li>- 5-membered heterocycles with one or more heteroatoms</li> <li>- 6-membered heterocycles with one or more heteroatoms</li> </ul>	<p><b>CALCULATIONS</b></p> <ul style="list-style-type: none"> <li>- Calculation of buffers pH</li> </ul> <p><b>BUFFER SYSTEMS</b></p> <ul style="list-style-type: none"> <li>- exercise: <i>Effect of acids and bases on pH of buffers</i></li> </ul>
6.	<p><b>1<sup>st</sup> REVISION TEST</b></p> <p><b>ORGANIC CHEMISTRY III</b></p> <ul style="list-style-type: none"> <li>- Biochemically and medicinally important derivatives of heterocyclic compounds (purines, pyrimidines, hormones, medicinal drug, dyes)</li> <li>- Heterocycles in drugs</li> </ul>	<p><b>CALCULATIONS</b></p> <ul style="list-style-type: none"> <li>- Balancing redox reactions</li> </ul> <p><b>PROPERTIES OF ORGANIC COMPOUNDS</b></p> <ul style="list-style-type: none"> <li>- exercise: <i>Preparation of esters of carboxylic acids</i></li> </ul>

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7.	<p><b>SACCHARIDES AND THEIR IMPORTANT DERIVATIVES</b></p> <ul style="list-style-type: none"> <li>- The relationship between structure and biological properties of saccharides</li> <li>- Biologically important monosaccharides and their derivatives</li> <li>- Disaccharides and polysaccharides</li> <li>- Complex saccharides and their importance</li> </ul>	<p><b>CALCULATIONS</b></p> <ul style="list-style-type: none"> <li>- Spectrophotometric calculations</li> </ul> <p><b>SPECTROPHOTOMETRY</b></p> <ul style="list-style-type: none"> <li>- exercise: <i>Spectrophotometric determination of copper with ammonia</i></li> </ul>
8.	<p><b>LIPIDS AND STEROIDS</b></p> <ul style="list-style-type: none"> <li>- Basic structure and classification of lipids</li> <li>- Fatty acids, their medical importance</li> <li>- Eicosanoids</li> <li>- Complex lipids</li> <li>- Steroids – classification, importance</li> </ul>	<p><b>SACCHARIDES</b></p> <ul style="list-style-type: none"> <li>- exercise: <i>Nitrochromium reaction</i></li> <li>- exercise: <i>Seliwanoff's reaction</i></li> <li>- exercise: <i>Tollen's reaction</i></li> <li>- exercise: <i>Tollen's reaction (silver mirror)</i></li> <li>- exercise: <i>Fehling's reaction</i></li> <li>- exercise: <i>Lugol test</i></li> </ul>
9.	<p><b>AMINO ACIDS, PEPTIDES</b></p> <ul style="list-style-type: none"> <li>- Classifications of amino acids, biochemical properties and their use in biochemistry</li> <li>- Amino acid derivatives and their biochemical significance</li> <li>- Peptides – structure, peptide bond, properties</li> <li>- Biochemically important peptides (e.g. glutathion)</li> </ul>	<p><b>LIPIDS</b></p> <ul style="list-style-type: none"> <li>- exercise: <i>Hydrolysis of neutral lipids by lipase</i></li> <li>- exercise: <i>Detection of double bonds in fatty acids</i></li> </ul>
10.	<p><b>PROTEINS</b></p> <ul style="list-style-type: none"> <li>- Spatial structure, classification, physical-chemical properties, biological and biomedical importance</li> <li>- Functional protein dispersity</li> <li>- Complex proteins and their function</li> <li>- Isolation methods and their use in medical practice</li> </ul>	<p><b>AMINO ACIDS</b></p> <ul style="list-style-type: none"> <li>- exercise: <i>Ninhydrin reaction</i></li> <li>- exercise: <i>Xanthoprotein reaction</i></li> <li>- exercise: <i>Foli's reaction</i></li> <li>- exercise: <i>Determination of unknown protein</i></li> </ul>
11.	<p><b>NUCLEIC ACIDS</b></p> <ul style="list-style-type: none"> <li>- Nucleotides and nucleosides</li> <li>- Biochemically important nucleotides with high energy hydrolysis</li> <li>- DNA and RNA - structure and biologic properties</li> <li>- Nucleic acid analysis techniques, their use in medical practice (restriction enzymes, PCR)</li> </ul>	<p><b>PROTEINS</b></p> <ul style="list-style-type: none"> <li>- exercise: <i>Reversible precipitation of proteins</i></li> <li>- exercise: <i>Irreversible precipitation of proteins</i></li> <li>- exercise: <i>Biuret reaction</i></li> </ul>
12.	<p><b>2<sup>nd</sup> REVISION TEST</b></p> <p><b>NATURAL COMPOUNDS, VITAMINS</b></p> <ul style="list-style-type: none"> <li>- Terpenes, alkaloids and flavonoids - structure physical-chemical properties, biological significance</li> <li>- General properties of vitamins - structure and their importance in biochemistry (e.g. coenzymes) and in medicine</li> </ul>	<p><b>NUCLEIC ACIDS I</b></p> <ul style="list-style-type: none"> <li>- exercise: <i>Isolation of nucleoproteins from eukaryotic cells</i></li> <li>- exercise: <i>Hydrolysis of nucleoprotein</i></li> </ul>
13.	<p><b>OXIDATIVE STRESS</b></p> <ul style="list-style-type: none"> <li>- Biological importance of free ions in biosphere</li> <li>- Formation of oxygen and nitrogen radicals</li> <li>- Antioxidants</li> </ul>	<p><b>NUCLEIC ACIDS II</b></p> <ul style="list-style-type: none"> <li>- exercise: <i>Detection of nucleic acids components</i></li> </ul> <p><b>3<sup>rd</sup> REVISION TEST – practical exercises, seminars</b></p>
14.	<p><b>MEMBRANES AND TRANSPORT</b></p> <ul style="list-style-type: none"> <li>- Structure of biological membranes</li> <li>- Membrane proteins</li> <li>- Transport of substances across biological membrane</li> <li>- Free diffusion, facilitated diffusion</li> <li>- Active transport, group translocation</li> </ul>	<p><b>EVALUATION OF STUDENTS' WORK</b></p> <ul style="list-style-type: none"> <li>- Final evaluation of practical exercises</li> <li>- Credit donation</li> </ul>